

Measurement Of K-shell Fluorescent Yield In Iridium - A Test Of Internal Conversion Theory

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 Research Experience for Undergraduates (REU), Summer 2004

Competing Forms of Nuclear De-excitation

- γ -ray emission
- Internal conversion

Internal conversion coefficients (ICCs):
 - Ratio of probabilities $\rightarrow \alpha = \frac{P_e}{P_\gamma}$

Internal Conversion (i)

$$\alpha = \alpha_K + \alpha_L + \alpha_M + \alpha_N + \dots$$

$$P_{K_x} = \omega_K \cdot P_{e_K} \quad \omega_K = \text{Fluorescent yield}$$

$$\alpha_K \omega_K = \frac{P_{K_x}}{P_\gamma}$$

$$N_\gamma = P_\gamma \cdot \epsilon_\gamma \quad N_{K_x} = P_{K_x} \cdot \epsilon_{K_x}$$

ICC, α_K

• 2004:
 - Nica [1] \rightarrow ICC for iridium

• Value from systematics [2], $\omega_K = 0.958(4) \rightarrow \alpha_K = 103.0(8)$

[1] N. Nica, Submitted to Phys. Rev. C
 [2] E. Schriedel and H. Jansen, Nucl. Instrum. Methods Phys. Res. Sect. A 368 127 (1996)

Why measure ICCs?

- General:
 - Essential role in analysis of nuclear decay schemes-
 - Spin, parity assignments
 - Transition rates
 - Branching ratios
- Specific:
 - Determine γ decay intensities (probabilities)-
 - Efficiency calibration of the ORTEC HPGe detector

Problems

- Systematic differences-
 - Different theoretical approaches
 - Theoretical and experimental (2-3%)
- Hole vs. No hole

• K-shell binding energy = 76.1-keV
 • $^{193m}\text{Ir} \rightarrow$ transition energy = 80.2-keV
 • $^{193}\text{Ir} \rightarrow$ transition energy = 129.4-keV

Experimental Method (i)

Source preparation:
 A-series:
 OsO₂ - Osmium (IV) Oxide

Series	Mass (mg)	Os Mass (mg)	Activity (kBq)	Plan. Fac.	(% Plan. Fac.)
A1	0.60	0.51	4.20	1.0000	1.0000
A3	1.52	1.30	10.70	0.9785	0.9800

C-series:
 OsCl₂·3H₂O - Osmium (III) Chloride Trihydrate

Series	Mass (mg)	Os Mass (mg)	Activity (kBq)	Plan. Fac.	(% Plan. Fac.)
C3	0.43	0.28	3.8	1.0000	1.0000

Experimental Method (ii)

Thermal Neutron Activation:

- Sources irradiated at the TRIGA reactor in the Texas A&M University Nuclear Science Center.
- Thermal neutron flux = $7 \times 10^{12} \text{ s}^{-1} \text{ cm}^{-2}$
- Time = 6 h

Significant isotopes produced (C3):

$^{194}\text{Os} \rightarrow ^{194}\text{Os}$	$A_1 = 8.5 \text{ kBq}$	$t_{1/2} = 93 \text{ d}$
$^{190}\text{Os} \rightarrow ^{190}\text{Os}$	$A_1 = 411 \text{ kBq}$	$t_{1/2} = 15.4 \text{ d}$
$\rightarrow ^{191m}\text{Os}$	$A_1 = (\text{assumed } \sim 0.1 \text{ kBq})$	$t_{1/2} = 13.1 \text{ h}$
$^{192}\text{Os} \rightarrow ^{192}\text{Os}$	$A_1 = 226 \text{ kBq}$	$t_{1/2} = 30.5 \text{ h}$

Sample spectra

Fluorescent Yield, ω_K

• Summer, 2004:
 - Montague, Nica
 \rightarrow experimental test of ICC

• Preliminary $\omega_K = 0.948(8)$ (calc. α_K for 129.4-keV trans. = 2.165 [3])
 • Survey value, $\omega_K = 0.958(4)$

[1] M.B. Truhlarovskaya, private communication

Corrections

• Coincidence summing:
 SummingCor - V.E. Jacob

n	Energy (keV)	Correction Factor
2 \rightarrow 1	82.427	1.00476
3 \rightarrow 1	129.431	1.00229
3 \rightarrow 2	47.050	1.03861
4 \rightarrow 3	41.846	1.01037

• Preliminary impurity analysis:

Nuclide	T-1/2	Activity (Bq)	% Total Actv.
^{193}Os	20.11 h	226238	0.2705
^{190}Os	83.6 d	8544.0	0.0102
^{194}Os	82.2 d	36972.8	0.0119
^{192}Os	2.72 d	177062.8	0.2117

... more to come ...

Conclusions

- This preliminary result for ω_K confirms our previously determined ^{193m}Ir α_K
- Confirms need to include 'hole' in theoretical calculations
- Future plans:
 - Continue analysis of ^{193}Os experimental data (impurities, additional spectra for decay curve analysis)
 - Complete similar experiments for cases with α_K sensitive to hole/no hole treatment (^{94}Nb)