

Homework Assignment #4

(*Due Date:* Thu, 10/19, 05:30 pm, in class; Show all your work for full/partial credit)

4.1 *Collision Kinematics* (cf. Quest. 4.9 in textbook) (1 pt.)

An electron ($m_e = 0.511 \text{ MeV}/c^2$) of kinetic energy 12.1 eV collides inelastically with a Xenon atom ($m_{Xe} = 131.2u$). Explain why essentially all of the incoming electron's kinetic energy is available to ionize the Xenon atom.

(*hint:* energy-momentum conservation in a heavy-light collision)

4.2 *Rutherford Scattering* (cf. Probs. 4.8+9 in textbook) (4 pts.)

A beam of α particles of 7 MeV kinetic energy is directed on a thin gold foil (Au nuclei have a radius of $\sim 7 \text{ fm}$ and charge $+79e$; the density of gold is $\rho_{Au}=19.3 \text{ g}/\text{cm}^3$).

- (a) If the rate of scattered α 's under a 30° angle is 100 per minute, what is the rate at (i) 10° (ii) 50° (iii) 90° (iv) 170° ?
- (b) If the incident energy is doubled, what is the rate of detected α 's under a 90° angle?
- (c) What is the minimal incident energy (in MeV) of α 's to cause deviations from the Rutherford formula?
At what angles would you expect the deviations to first occur?
- (d) If the original gold foil is replaced by an aluminum foil ($\rho_{Al}=2.7 \text{ g}/\text{cm}^3$, $Z_{Al}=13$) of identical thickness, what is the rate of scattered α 's under 30° ?

4.3 *Hydrogen Transitions* (cf. Quest. 3.11 in textbook) (1 pts.)

Consider 3 transitions of electronic orbits in hydrogen characterized by their initial (n_i) and final (n_f) quantum numbers as follows:

- (i) $n_i=1$ and $n_f=4$ (ii) $n_i=5$ and $n_f=3$ (iii) $n_i=3$ and $n_f=7$.

Which transition involves the shortest wavelength and which transitions are absorption processes?

4.4 *Bohr Model of the Atom* (cf. Probs. 4.23+24 in textbook) (4 pts.)

A hydrogen model is initially at rest and in its third excited state ($n_i=4$).

- (a) Calculate the radius of the orbit (in nm), as well as the electron's kinetic, potential and total energy (all in eV).
- (b) The atom de-excites into the ground state. Calculate the wavelength of the emitted photon (in nm) as well as the recoil momentum (in eV/c) and kinetic energy (in eV) of the atom.
(*hint:* momentum conservation)